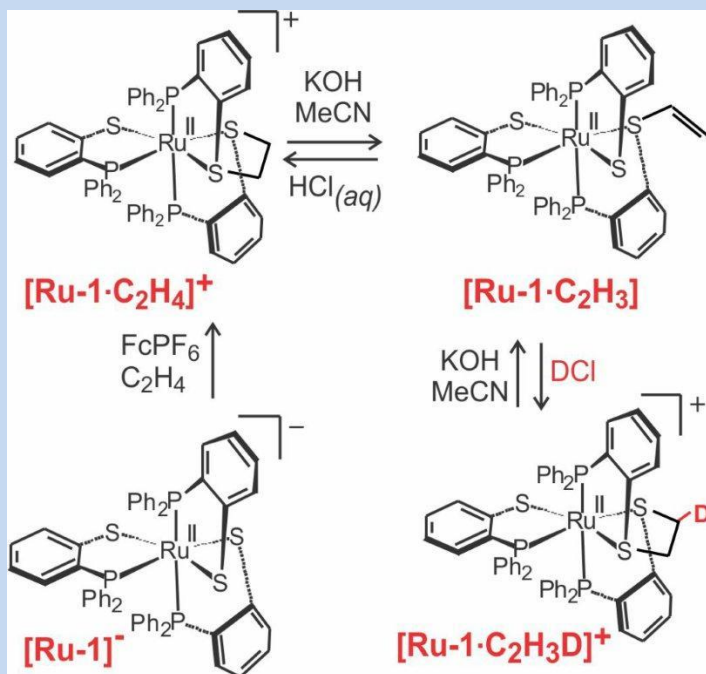


Electrochemically-Controlled Alkene Binding Affinity of Surface-Immobilized Metal-Thiolate Complexes

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Deprotonation of $[\text{Ru-1-C}_2\text{H}_4]^+$ with potassium hydroxide yields the vinyl metallosulfonium derivative $[\text{Ru-1-C}_2\text{H}_3]$ in acetonitrile via an elimination reaction. Acidification of $[\text{Ru-1-C}_2\text{H}_3]$ with HCl or HBr regenerates $[\text{Ru-1-C}_2\text{H}_4]^+$. The reversible deprotonation proceeds selectively at the carbon alpha to the sulfur trans to phosphorus. Deuteration of $[\text{Ru-1-C}_2\text{H}_3]$ with DCl yields the mono-deuterated dithioether complex $[\text{Ru-1-C}_2\text{H}_3\text{D}]^+$, which yields the deuterium free product $[\text{Ru-1-C}_2\text{H}_3]$ upon deprotonation. We look to exploit this for selective deuteration of alkenes in the future.



A film of $[\text{Re-1}]^+$ drop-coated across gold electrodes exhibits an increase in resistance upon exposure to ethylene. This is due to selective ethylene binding to form $[\text{Re-1-C}_2\text{H}_4]^+$, which exhibits higher resistance due to less electron delocalization in the metal-thiolate complex. The binding is reversible, leading to release of ethylene in the presence of nitrogen and a return to the original resistance.

