

# Improved lines positions for the (1,1) band of the $b\ 1\Sigma^+$ - $X\ 3\Sigma^-$ transition of $O_2$

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## INTRODUCTION

Molecular oxygen,  $O_2$ , is an essential life-supporting component in our atmosphere and oceans. Both the concentration and the distribution of gas-phase oxygen in the atmosphere are of interest to many science disciplines, including planetary scientists. We report improved line positions for the (1,1) vibrational band of the  $b\ 1\Sigma^+$  -  $X\ 3\Sigma^-$  transition of  $O_2$ , recorded using intracavity laser absorption spectroscopy.

## EXPERIMENTAL METHOD

The absorption spectra were recorded using intra-cavity laser spectroscopy (ILS). A plasma discharge was used to enhance absorption from  $v=1$  in the ground state of oxygen.

## RESULTS AND DISCUSSION

A portion of the spectrum with  $^R R$ - and  $^R Q$ -branch spectral features is shown in Figure 1. High lines of the (0,0) band of this transition can be seen in the spectral region of the (1,1) band, and the  $^P P(29)$  and  $^P Q(28)$  lines of the (0,0) band are labeled in Figure 1. A total of 66 lines with  $J$ 's up to 36 were fitted in this analysis. Molecular constants obtained from the fit are presented in Table 1 and compared with previous literature values. In summary, improved lines positions for the (1,1) band of the  $b\ 1\Sigma^+$  -  $X\ 3\Sigma^-$  transition of  $O_2$ . Additionally, we report a new method of producing vibrationally hot molecules for use in absorption spectroscopy of stable gas phase molecules.

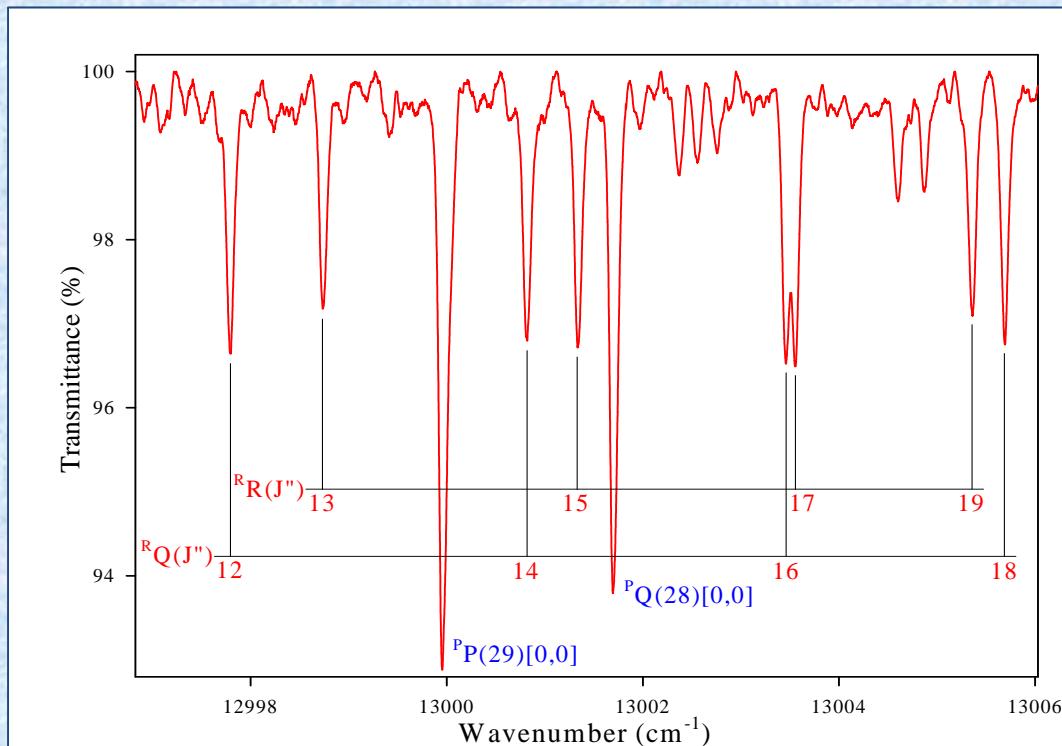


Figure 1. A portion of the the (1-1) band of the  $b\ 1\Sigma^+$  -  $X\ 3\Sigma^-$  transition of  $O_2$  (in  $\text{cm}^{-1}$ ):  $P_{\text{discharge}}=4.70$  torr,  $P_{\text{background}}=4.26$  torr, and  $t_g=200$   $\mu\text{sec}$ .

Table 1. Molecular Constants for the  $v=1$  of the  $b\ 1\Sigma^+$  state of  $O_2$  (in  $\text{cm}^{-1}$ ). The constants of the  $X\ 3\Sigma^-$   $v=1$  state were held fixed to the values of Rouillé *et al.* [16].

Molecular Constant	$E_1$	$B_1$	$D_1 \times 10^5$
This Work	14526.98928(84)	1.3729747(37)	0.54157(31)
Ref. 2	14525.6602	1.373054	0.5409